

## **Materials Science and Engineering**

**Evaluation Test** 

**TOPIC 2. STRUCTURE OF MATERIALS** 

**Important**: Mark the right answer with a X. The correct answers will mark + 1 points while the incorrect answers will mark as -0.33 points. Non answered questions will not mark nor positively nor negatively. The resulting mark will not be smaller than 0 in any case. There is only one correct answer per question. Good luck!

A metal with a FCC structure has a lattice parameter of 4.08 Å and density of $\rho$ = 19.3		
g/cm <sup>3</sup> . Calculate the atomic weight of the metal ( $N_A = 6.023 \times 10^{23}$ atoms/mol).		
197.4 g/mol		
394.7 g/mol		
56.4 g/mol		
2x10 <sup>23</sup> g/mol		

The atomic radius of an atom that is present in a BCC unit cell having a lattice		
parameter of 2.87 Å is:		
2.48 Å		
1.01 Å		
1.43 Å		
1.24 Å		

For a screw dislocation		
	The burgers vector is perpendicular to the dislocation line and parallel to the	
	direction of dislocation movement.	
	An extra half-plane of atoms is introduced in the lattice.	
	The burgers vector is parallel to the dislocation line and perpendicular to the	
	direction of dislocation movement.	
	The movement occurs easier in non closed- packed directions.	

Diffusion occurs faster:		
	In materials with high melting point.	
	By vacancy diffusion than by interstitial diffusion.	
	In the BCC structure rather that in the FCC structure.	
	Along the volume (crystal lattice) than along the grain boundaries.	

The Miller Indices for the following plane are:		
	(312)	
	(132)	
	[ <u>3</u> ]	
	(1 3 2)	

