## uc3m Universidad Carlos III de Madrid

OpenCourseWare (2023)

## **CHEMISTRY II**

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## EXERCISES OF ORGANIC COMPOUNDS WITH OTHER FUNCTIONALITIES



**Exercise 1.** Propose the Grignard reagent and the carbonyl compound required to prepare the following alcohols:

a) b) c) 
$$CH_3$$
  $OH$   $CH_3$   $OH$   $OH$ 

**Exercise 2.** Indicate the products which result from reaction of 1-methylcyclohexanol with the following reagents: a) HBr, b)  $H_2SO_4$ , and c)  $Na_2Cr_2O_7$ .

**Exercise 3.** Indicate how you would prepare the following ethers using a Williamson synthesis:

- a) Benzyl isopropyl ether
- b) Ethyl 2,2-dimethyl propyl ether

**Exercise 4.** Which products would be obtained from the following ether cleavage reactions?

a) 
$$O$$
  $CH_3$   $H_2O$ 

b) 
$$CH_3$$
  $CF_3CO_2H$  ?

c) 
$$H_3C \xrightarrow{CH_3} O \xrightarrow{CH_3} H_2O$$
 ?

**Exercise 5.** Predict the products of the reaction of phenylacetaldehyde with the following reagents:

- a) NaBH<sub>4</sub> followed by H<sub>3</sub>O<sup>+</sup>
- b) Dess-Martin reagent

**Exercise 6.** Indicate the reagents required to synthesize the following compounds by using a Grignard reaction:

- a) 1-Butanol
- b) 1-Phenylcyclohexanol
- c) Diphenylmethanol

**Exercise 7.** Justify which of these two acids, lactic acid and acetic acid, will be stronger.

**Exercise 8.** Rank the following compounds in order of increasing acidity: *p*-Nitrobenzoic acid, acetic acid, and benzoic acid.

**Exercise 9.** Indicate which method, Grignard carboxylation or nitrile hydrolysis, is more appropriated to carry out the following reactions:

b) 
$$H_3C$$
 $CH_3$ 
 $H_3C$ 
 $CH_3$ 
 $CH_3$ 

**Exercise 10.** How might you prepare the following ester from the corresponding carboxylic acid?

**Exercise 11.** Explain the decreasing  $pK_a$  values of the following (protonated) amines:

$$H_3C$$
 $NH_2$ 
 $H_2C$ 
 $NH_2$ 
 $HC$ 
 $NH_2$ 
 $NH$ 

Exercise 12. What product would you expect from Hofmann elimination of the following amine?

**Exercise 13.** Indicate the precursors used to prepare the following amines by means of reductive amination reaction.

## **IMAGE CREDITS**

Images of all exercises were made by authors.